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Length Of The Wire, 15.1 Cm), And T Is The Previously Calculated Temperature Rise, 2.77252 C . Combining  
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Change In Temperature Of The Solution = T Final T  
Initial In This Experiment Mass Of The Solution Will Be  
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Density (assumed To Be 0.997 G ML<sup>-1</sup>) Heat Of  
Neutralization Is The Same Value As Q Soln But With  
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Consumption Calorimetry, William Parker: 2016 ...Of  
The ASTM E84 Test Method. The Work Had Been Done  
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 $(0.875 \text{ Mol})(1350 \text{ KJ/mol}) = 1180 \text{ KJ} = 118000 \text{ J}$  Plug This Value Into Our Equation And Then Rearrange To Find The Change In The  
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 1. Using The Graduated Cylinder, Obtain 50 ML Of Water And Carefully Pour It Into The Soda Can.  
 2. Determine The Mass Of Water And The Can. Record The Mass Of Water In The Data Table (hint: Density Of Water = 1 G/mL).  
 3. Hold The

Paper Clip Horizontally And Bend The Outer End Upwards Until Mar 5th, 2024.

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