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Standard States The Thermodynamic Standard State Of A Substance Is Its Most Stable Pure Form Under Standard Pressure (1 Bar) And Some Speci C Temperature (298 K Unless Otherwise Speci Ed). I For A Pure Substance In The Liquid Or Solid Phase, The Standard State Is Th Apr 12th, 2024

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Thermochemistry © 2009, Prentice-Hall, Inc. Energy •Energy Is The Ability To Do Work Or Transfer Heat. -En Feb 5th, 2024

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162 CHAPTER 6: THERMOCHEMISTRY To Convert The Answer To Joules, We Write: $101.3 \text{ J} \cdot 0.18 \text{ L} \cdot \text{Atm} \cdot 1 \text{ L} \cdot \text{Atm}^{-1} \cdot \text{m}^3 = \cdot \text{W} - 18 \text{ J}$ 6.17 An Expansion Implies An Increase In Volume, Therefore W Must Be -325 J (see The Defining Equation For Pressure-volume Work.) If The System Absorbs Heat, Q Must Be $+127 \text{ J}$. The Change In Energy (internal Jan 17th, 2024

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To Thesum Ofenergyin Form Heat Exchanged Between System And Surroundings (q) And The Work Done On The System (w). $\Delta E_{\text{system}} = Q + W$ But The Laws Of Thermodynamics Are Far More Profound Than Simply Describing The Efficiency Of Steam Engines. T Apr 4th, 2024

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