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Probably Does Not Dissolve In Water. For Example, The Electronegativity Of Mercury Is 1.9. The Electronegativity Of Sulfur Is 2.5. The Difference In These Two Electronegativities Is Small, Only 0.6. Therefore, You Can Predict That Mercury(II) S Mar 4th, 2024

Factors That Affect Solubility

Increases. Suppose You Had A Spoonful Of Sugar Granules And The Same Amount Of Sugar Pressed Into A Cube. The Spoonful Of Sugar Has A Greater Total Surface Area Than The Cube, Because Each Granule In The Spoonful Contacts The Solvent Separately. So, The Spoonful Of Sugar Will Dissolve At A Faster Rate Than The Sugar Cube. Mar 1th, 2024

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