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CHE 31. INTRODUCTION TO CHEMICAL ENGINEERING CALCULATION Example 9-4. Absorption Of SO₂ The

Molar Density Of The Feed Gas May Be Determined From The Formula $(\text{mol/L}) = \frac{P}{RT}$ Where P And T Are The Absolute Pressure And Temperature Of The Gas. The Following Data Are Taken For The Feed Gas: T=750F, P=150 Psig, H=210 Mm, R=82.4 And For The Outlet Gas, R=11.6. Determine The (a) Mass Fraction Of SO₂

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Presentation. ... Mass Of A Compound That Comes

From Each Of The Elements In The Compound By Using

This Equation: % Element = $\frac{(\text{number Of Atoms})(\text{atomic Weight})}{(\text{FW Of The Compound})} \times 100$. Jan 9th,

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Formulas The Compound Para-aminobenzoic Acid (you

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