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T? As A Sample Application, We Can Assume The Ideal Gas Equation Of State. $V = RT/P \Rightarrow (\partial V/\partial T)P = R/P$. Substitution $\Rightarrow \Delta S = C P Ln(T 2/T 1) - R Ln(P 2/P 1)$ Applying $\Delta S = 0 \Rightarrow Ln(T 2/T 1) = R/Cp Ln(P 2/P 1) \Rightarrow (T 2/T 1) = (P 2/P 1) (C P-C V)/C PTTPV P Mar 4th, 2024$

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