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Ch. 7: Chemical Reactions And Chemical Quantities

Ch. 7: Chemical Reactions And Chemical Quantities Dr.

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Quantities1-carbon 2-oxygen Thus One Mole Of CO₂
Contains Three Times Avogadro's # Of Atoms.
Example: How Many Atoms Are In 2.12 Mole Of
Propane (C₃H₈)? Know: # Of Moles = 2.12 Mole C₃H₈
H₈ 1 Mole C₃H₈ = 6.02 X 10²³ Molecules C₃H₈ 1
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