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And Products Remain Constant With Time. Consider
The Following Reaction: H 2O + CO Æ H 2 + CO 2
Suppose You Were To Start The Reaction With Some
Amount Of Each Reactant (and No H Mar 1th, 2024

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F Pivot =(m B +m 1 +m 2)g F Pivot -m B G -N B,1 -N B,2 =0 Worked Example: Solution Pivot Force: Lever Law: Pivot F =(m B +m 1 +m 2)g =(2.0 Kg +0.3kg +0.6 Kg)(9.8 M·s-2) =28.4 N D 1 M 1 =d 2 M 2 D2 =d1m1 / M2 =(0.4 M)(0.3 Kg / 0.6 Kg) =0.2 M Generalized Lever Law , , 1 11 22, 2,  $\bot$   $\bot$  =+ =+ FF F FF F & & GG G GGG Mar 1th, 2024

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A) Keq 1 D) Keq Cannot Be Determined. 6 Concentration And Solubility Of Gas The Solubility Of CO2 Gas In Water Is 0.240 G Per 100 MI At A Pressure Of 1.00 Atm And 10.0°C. Apr 1th, 2024

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