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Calculating PH And POH Worksheet 4) - A Solution Is Created By Measuring 3.60 X 10³ Moles Of NaOH And 5.95 X 10⁻⁴ Moles Of HCl Into A Container And Then Water Is Added Until The Final Volume Is 1.00 L. What Is The PH Of This Solution? Since There Is Both Acid And Base We Will Assume A 1 Mole Acid:1 Mole Base Ratio Of Neutralization. Mar 4th, 2024 PH And POH - Ms. Mogck's Classroom - Ms. Mogck's Classroom KEY Chemistry: PH And POH Calculations Part 1: Fill In The Missing Information In The Table Below. PH [H³ O¹⁺] POH [OH¹⁻] ACID Or BASE? 3.78 1.66 X 10⁻⁴ M 10.22 6.03 X 10⁻¹¹ M Acid 3.41 3.89 X 10⁻⁴ M 10.59 2.57 X 10⁻¹¹ M Acid 8.81 1.55 X 10⁻⁹ M 5.19 6.46 X 10⁻⁶ M Base 8.69 2.04 X 10⁻⁹ M 5.31 4.88 X 10⁻⁶ M Base 8.46 3.47 X 10⁻⁹ M 5.54 2.88 X 10⁻⁶ M Base Mar 4th, 2024 Basic Concepts Of Chemistry And Chemical Calculations Elements. For Example, Sodium Is A Shiny Metal, And Chlorine Is An Irritating Gas. But The Compound Formed From These Two Elements, Sodium Chloride, Shows Different Characteristics As It Is A Crystalline Solid, Vital For Biological Functions. Physical Classification Solid Mixtures Ice Water Water Vapour Homogeneous Feb 2th, 2024.

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