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(A) 0.0500 Mole (B) 0.0625 Mole (C) 0.125 Mole Apr  
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Should Be. A.  $\text{C}_4\text{H}_{10} / \text{O}_2$  B.  $\text{O}_2 / \text{CO}_2$  C.  $\text{O}_2 / \text{H}_2\text{O}$   
D.  $\text{C}_4\text{H}_{10} / \text{CO}_2$  E.  $\text{C}_4\text{H}_{10} / \text{H}_2\text{O}$   
2. Given The

Following Equation:  $2 \text{KClO}_3 \rightarrow 2 \text{KCl} + 3 \text{O}_2$  A. How Many Moles Of  $\text{O}_2$  Can Be Produced By ...File Size: 80KBPage Count: 2 Apr 5th, 2024

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