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Chatelier's Principle, Catalysts Changes In Equilibria.
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VIII) Strengths Of Covalent Bonds For Ionic Compounds
LE Is An Indication Of The Strength Of Attraction Of The
Ions. For Covalent Bonds The Strength Is Measured As
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Example 9-4. Absorption Of SO₂ The Molar Density Of The Feed Gas May Be Determined From The Formula $(\text{mol/L}) = 12.2P(\text{atm})T^{-1}(\text{K})$ Where P And T Are The Absolute Pressure And Temperature Of The Gas. The Following Data Are Taken For The Feed Gas: T=750F, P=150 Psig, H=210 Mm, R=82.4 And For The Outlet Gas, R=11.6. Determine The (a) Mass Fraction Of SO₂
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