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= NRTIdeal Gas Law Worksheet PV = NRT Use The Ideal Gas Law, “PerV-nRT”, And The Universal Gas Constant R = 0.0821 L*atm To Solve The Following Problems: K*mol If Pressure Is Needed In KPa Then Convert By Multiplying By 101.3kPa / 1atm To Get R =8.31 KPa*L / (K*mole) Mar 2th, 2024.
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