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1.1 Rate Of Reaction And Dependence On Temperature We Will Once Again Look At  
The Formation Of Ammonia (NH<sub>3</sub>) From Nitrogen And Hydrogen (see Section  
Chemical Equilibrium Of The Thermodynamics Chapter). This Reaction Follows The  
Equation:  $N_2 + 3H_2 \rightleftharpoons 2NH_3$  (1)  $H_0 = 92 \text{ kJ Mol}^{-1}$   $S_0 = 192 \text{ J Mol}^{-1} \text{ K}^{-1}$  Feb 16th,  
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