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Example Problems 1. Consider The Galvanic Cell Based On The Reaction $\text{Al}^{3+} + \text{Mg} \rightarrow \text{Al} + \text{Mg}^{2+}$ The Half

Reactions Are $\text{Al} \rightarrow \text{Al}^{3+} + 3\text{e}^{-}$ $E^{\circ} = 1.66 \text{ V}$ $\text{Mg} \rightarrow \text{Mg}^{2+} + 2\text{e}^{-}$ $E^{\circ} = 2.37 \text{ V}$

Give The Balanced Cell Reaction And Calculate E° For The Cell. (See Work On Page 845) Solution: 0.71 V

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