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Chemistry 11 Stoichiometry Review Worksheet Name: ___KEY___ 33. $\text{KClO}_4 \rightarrow 2\text{O}_2 + \text{KCl}$ Actual yield = 1.45g

Theory yield = $5.95\text{g KClO}_4 \times \frac{1\text{mol}}{122.55\text{g}} \times 2 \times 32.00\text{g/mol} = 3.11\text{g O}_2$
%Yield = $\frac{1.45\text{g}}{3.11\text{g}} \times 100\% = 46.6\%$ 34. $4\text{FeCO}_3 + \text{O}_2 \rightarrow 2\text{Fe}_2\text{O}_3 + 4\text{CO}_2$

Percent purity = 62.8%? g pure FeCO_3 Feb 4th, 2024
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