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4 • ? H<sub>2</sub>O Weighing 24.95 g Is Heated Until All The Water Is Removed. The Anhydrate Weighs 15.95 g. What Is The Value Of ? In The Hydrated Salt? (MM CuSO<sub>4</sub>, 159.5 g/mol) A. 2 B. 3 C. 4 D. 5 27. The Simplest Formula For A Hydrocarbon That Is 20.0 Percent Hydrogen By M Apr 6th, 2024

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Percent Composition And Empirical Formulas 8) What Is The Mass Percent Of Each Element In The Following Compounds? A. CaCl<sub>2</sub> B. Fe<sub>2</sub>O<sub>3</sub> C. C<sub>6</sub>H<sub>10</sub>S<sub>2</sub>O<sub>9</sub> 9) Calculate The Empirical Formulas Of Compounds Containing The Following Percentages Of Elements. Use The Molar Mass To Calc Feb 6th, 2024

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(Show All Work) 1. MgO 5. Ca(OH)<sub>2</sub> 2. NaHCO<sub>3</sub> 6. CH<sub>4</sub> 3. C<sub>6</sub>H<sub>12</sub>O<sub>6</sub> 7. NH<sub>3</sub> 4.  
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— Using The Molar Mass Of Each Substance, Convert The Following Quantities. A. 10.0 Mol Cr 520 G F. Mole On Grams, Grams On Moles Conversion Worksheet Units - 4 Moles And Stoichiometry Site 1 Of 2 Moles On Grams, Grams On Moles Conversion Worksheet What Are The Molecularthesis Of The Following Compounds? 1) NaOH 2) H 3PO 4 Apr 8th, 2024

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