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Answer On Question #60099, Chemistry / Inorganic Chemistry 1. 1.00g Of Carbon Is Combusted In A Limited Supply Of Pure Oxygen. 0.50g Of The Carbon Combusts To Form CO_2 And 0.50 G Of The Carbon Combusts To Form CO . The Resultant Mixture Of CO_2 And CO Is Passed Through Excess Na Mar 15th, 2024

Answer On Question #58182 Chemistry Organic Chemistry ...

2) Aniline And Acetanilide Stand In A One-to-one Molar Relationship, So 0.0322 Mol Of Aniline Produces 0.0322 Mol Of Acetanilide. The Amount Of Acetic Anhydride Does Not Play A Role, Since It Is In Excess. Then, $n(\text{N}) = 0.0322 \text{ mol aniline} \times \frac{1 \text{ mol Acetanilide}}{1 \text{ mol Aniline}}$. We Find The Molar Mass Of Acetanilide: $(12 \times 8) + (1 \times 9) + (14 \times 1) + (16 \times 1) + 135 = 178 \text{ g/mol}$.
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