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1.1 Rate Of Reaction And Dependence On Temperature We Will Once Again Look At The Formation Of Ammonia (NH_3) From Nitrogen And Hydrogen (see Section Chemical Equilibrium Of The Thermodynamics Chapter). This Reaction Follows The Equation: $\text{N}_2 + 3\text{H}_2 \rightleftharpoons 2\text{NH}_3$ (1) $H^\circ = 92 \text{ KJ Mol}^{-1}$ $S^\circ = 192 \text{ J Mol}^{-1} \text{ K}^{-1}$ Jan 6th, 2024

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