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87 The Total Number Of Moles, N, Is The Sum Of The Number Of Moles Of All The Components Of The Mixture: $N=N1+N2+\ldots$ It Follows That The Sum Of All The Mole Fractions Of The Mixture Must Also Equal 1. $X1+X2+\ldots=1$ The Mass Of Component I In A Mixture Is The Product Of The Number Of Moles Of I And Its

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The Enthalpy Of Combustion Of The Fuel. Jan 8th, 2024

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